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PPSC CHEMISTRY PAPER LECTURER 2015 MCQS

1. Enzymes belong to which class of compound-----Proteins----- **right answer)**
2. The helical structure of protein is stabilized by -----Hydrogen-----bond. **right answer)**
3. Enzymes have...Polypeptide.....unit
 - A)POLYpeptide **right answer)**
 - B)HYDROCARBON
4. Function of enzymes in biological system is.....
Catalyze biochemical action **right answer)**
5. Night blindness is produce due to deficiency of ...
.....VITAMIN A **right answer)**
6. Zwitter ion exist in which of the following
 - a)Alanine **right answer)**
 - b)Glycine HCL
 - c) Both a and b
 - d)None
7. IR determine -----bond
 - a)kind of bond nature of bond **right answer)**
 - c)hydrocarbon
8. Marsh gas is
...CH4 **right answer)**
9. Solvent used in IR Spectroscopy is.....
 - a)CCL4 **right answer)**
 - B)CH3OH
 - C)WATER

D)CH₃CH₂OH

10. 1700-----1760 PEAK In IR Spectroscopy is.....

...Carbonyl group **right answer)**

11. NMR use -----source of radiation

a)UV

B)Radio waves **right answer)**

c)micro waves

d)visible rays

12. Existence of nucleus is confirmed by.....

Millikian method or

Alpha scattering by thin metal foil **right answer)**

13. What is faze about electron (Not conform but it should be d) a) Electron is a particle

b) Wave is associated with electron

c) When electron jump from orbital its energy c..

d) oriented by magnetic field

14. Which one is occur in whole number always

a) Atomic number **right answer)**

b) Molecular weight

c) Equivalent weight

15. A metal has an electronic configuration is M+2 has 2,8,14 and its atomic weight is 56.the number of neutron is

36) **right answer)**

16. Least soluble in Water Is

a. Ethano

b. I b. Phenol

c. c. Carboxylic acid

d. d. benzene **right answer)**

17. The EAN Value of Ni(CO)₄ is

a)36 **right answer)**

b)54

18. The ration of C:Si in their neutron ration with atoic weight having 12 and 28 respectively (3:7)

right answer)

19. Phenolphthalein is colourless in acidic medium due to ___

__ Benzenoid form (right answer)

20. The weakest base is

- a) H⁻
- b) Cl⁻ (right answer)
- c) HCO₃⁻
- d) OH⁻

21. Which compound decolorize KMnO₄ solution but not give test with AgNO₃ solution

- a) Ethylene (right answer)
- b) Acetylene
- c) Ethane

22. Which one of the following does not give Lassaignes test for nitrogen

- A. Phenylhydrazine
- B. Glycine
- C. Urea
- D. Hydroxylamine (right answer)

23. which one the following compound does not explode:

- a) trinitro toluene
- b) 2-amine toluene (I think this is not conform)

24. Strongest oxidizing agent is supported by-----

- a. Low dissociation energy
- b. High electron affinity (right answer)
- c. Low ionization energy

25. the order of oxidizing agent is I < Br < Cl < F

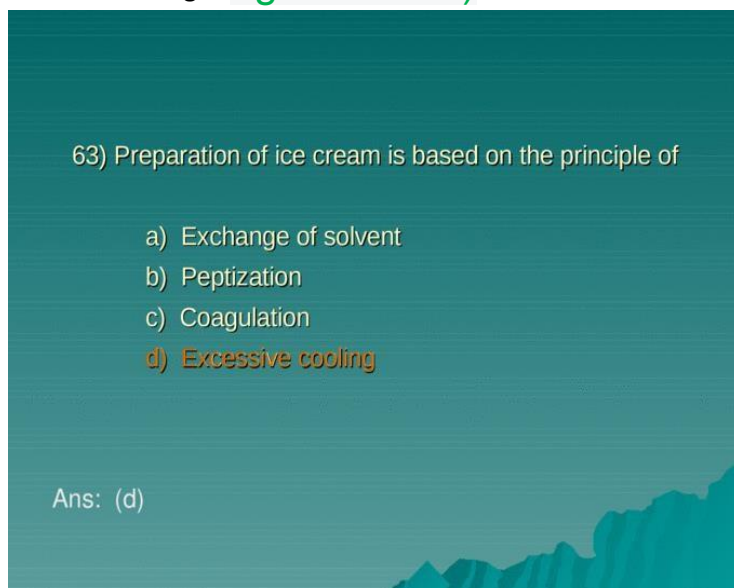
26. inversion of cane sugar into glucose and fructose is ?(not conform)

- a) slow process fast process
- b) spontaneous process instantaneous process

27. rate of chemical reaction is-----when it proceed

- a. decrease (right answer)

- b. increase
28. when reactants are added then rate of reaction is (not conform)
decrease increase
29. which one of the following metal has highest electrical conductivity
- A. gold
 - B. silver (right answer)
 - C. aluminium
 - D. copper
30. preparation of ice cream is
- a) colloids
 - b) excessive cooling (right answer)



- c) coagulation
 - d) peptization
31. colloids are example of -----equilibria
- a. homogenous equilibria
 - A. hetrogenous equilibria (right answer)
 - b. suspension
 - c. none of these
32. pure water is nonconductor of electricity
- A. water is neutral
 - B. water is almost unionized (right answer)

- B. water is completely ionized
33. which one of the following aqueous solution conductor of electricity
- C. Glucose
- D. HCl (right answer)
34. Which is correct statement (not confirm my answer is c)
- A. Catalyst increase the rate of forward reaction
- B. Catalyst increase the rate of backward reaction
- C. Catalyst influence on the rate of forward reaction and backward reaction equally
- D. Catalyst decrease the rate of forward reaction
35. Pick a right statement:

Order of reaction is in fraction (right answer)

36. The unit for formation of ammonia from $N_2 + 3H_2 \rightarrow 2NH_3$

Mol².lit⁻² (right answer)

37. The unit of active mass is-----

- A. Gram atom per litre (right answer)
- B. Gram molecule per litre
- C. Gram litre

38. colloidal particles are
- a) negative
- b) positive
- c) neutral (my answer is neutral not confirm)
- d) may be positive or negative

39. who classify element first

- a) lother meyer (right answer)
- b) doberienner
- c) newlands
- d) mendeleevs

40. molecularity of reaction is always inwhole number (right answer)

41. the chemical which destroy ozone layer is $CFCl_3$ (right answer)

42. physisorption occur at -----temperature low (right answer)

43. if dispersion medium is gas its example is

Aerosol (right answer)

44. pressure vs temperature equation is called (not conform answer)

- A. Langmuir equation
 - B. Fredluck equation
 - C. Bet equation
 - D. None of these
45. Resonating structure cannot be separated (right answer)
46. Nickel is use in the hydrogenation of vegetable ghee (right answer)
47. Colloids shows Brownian movement (right answer)
48. Methyl group is Ortho and p-directiong bcz.....
(Not conform my answer is all ob above)
- a) E donating effect of methyl group inductively
 - b) E donating effect of methyl group resonatingly
 - c) E donating effect of methyl group hyperconjugation
 - d) All of above
49. Chloro group is ortho and para but electron withdrawing (right answer)
50. For a Reversible reaction to occur
- A. Glass vessel
 - B. Closed vessel (right answer)
 - C. Open vessel
51. In benzene, there are -----type of carbon atom
- A. One type (right answer)
 - B. Two type
 - C. Three type
 - D. Four type
52. In benzene, the number of pi electron is
- A. 3 e
 - B. 4e
 - C. 6e (right answer)
 - D. 4e
53. In benzene , all carbon atoms are-----hybridization
- A. Sp3

- B. sp^2 (right answer)
C. sp
D. None
54. According to Le Chatelier principle in a reversible reaction between solid and liquid, the amount of heat added
(Not conform answer)
A. Decrease the concentration of solid
B. increase the concentration of solid
C. Decrease the concentration of liquid
D. Decrease the concentration of liquid
55. The fastest reaction occur in which of the following:
A. Rusting of iron
B. Burning of coal
C. The $AgCl$ formation by the reaction of $AgNO_3$ and $NaCl$ (right answer)
56. The amount of energy required to separate nucleons is called
A. Ionization energy
B. Binding energy (right answer)
57. The enthalpy of heat of formation of sodium ion from its atom is known as
A. Enthalpy of atomization
B. First ionization energy (not conform)
C. Enthalpy of combustion
58. The brown vapour is formed of which element when MnO_2 and conc. H_2SO_4 is added in them
A. NO_2 (right answer)
B. Cl_2
C. Br_2
D. I_2
59. Stability of nucleus is due to
A. Proton and neutron (right answer)
B. Proton and electron
C. Neutron and electron
60. Difference between particle mass and real mass is called..... Mass defect (right answer)
61. A gas has a volume of 2 litre at S.T.P at constant pressure, the new volume is 4 litre at which temperature:

- A. 50c (right answer)
B. 100c
62. The difference between the crystalloids and colloids
- A. Particle size (right answer)
B. Colloidal size
C. Diffusion
D. True solution
63. The gas which shows deviation from ideal behavior is:
- A. Helium
B. Hydrogen
C. Ammonia (right answer)
D. Trichloro methane
64. Some substance shows scattering of light but they can pass through ordinary filter paper is.....
- A. Colloids (right answer)
B. Suspension
C. True solution
D. None
65. In the lime kiln , $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ proceed to complete because.....
- A. CO_2 ESCAPPE
B. CaO remove
C. Low Temperature
D. High Pressure
66. Which one of the following equation is not affected by change in pressure
- A. $\frac{1}{2}\text{N}_2 + \frac{1}{2}\text{O}_2 \rightarrow \text{NO}$ (right answer)
B. $\text{P}_2\text{O}_5 \rightarrow \text{P}_2\text{O} + \text{O}_2$
C. $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
67. A Reaction approaches go farthest completions
- A. $K=10^2$
B. $K=10^1$
C. $K=10$
D. $K=10^{-2}$
68. Isothermal thermal system is one in which

$\Delta T = 0$ (right answer)

$\Delta E = W$

$V = 0$

$E = 0$

69. Temperature of the system decreases in adiabatic expansion (right answer)

adiabatic compression

70. System in which no change in temperature is Isochoric isobaric

isothermic (right answer)

71. Cyclic structure of benzene is given by

Kekule (right answer)

Doberiner

72. Which is least soluble in water

A. Ethanol

B. Benzene (right answer)

C. Benzoic acid

D. Phenol

73. Heterogenous catalysis is based on

a. Adsorption (right answer)

A. Absorption B.

Sorption

C. Desorption

74. Entropy of the universe is

Always increasing (right answer)

Decreasing

Constant

75. Which indicator is not used in conical flask

A. Self indicator

B. External indicator (right answer)

C. Internal indicator

D. Mixed indicator

76. Toluene is prepared from benzene

- A. Friedel craft reaction (right answer)
- B. Perkin reaction
- C. Wurtz reaction

77. Catalyst used in Friedel craft alkylation is

- A. $AlCl_3$ (right answer)
- B. $BFeCl_3$

78. Order of reaction is.....

Can be determined from equation

Can be determined from experiment (right answer)

79. Oxidation number of P in KH_2PO_2 is

- +1 (right answer)
- +3
- +5
- +2

80. Molten KCl conduct electricity because of

Free ions (right answer)

Free electron

81. which individual units lose identity

- a. complex salts
- b. simple salts

c. compound salts (right answer)

82. When a multidentate ligand is surrounded it is known as

- Coordination sphere
- Coordination complex

Chelates (right answer)

83. Group no. represents

Same electronic configuration same number of electron in valance

shell (right answer)

84. Entropy decreases in melting of ice rusting of iron