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# PPSC, FPSC (Lecturer- Chemistry) Solved Mcqs Preparation Notes-Part-4 On doc4shares.com

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151. QUESTION: Potassium was isolated in \_\_\_\_\_.

- A. 1707
- B. 1807
- C. 1907
- D. 1007

**CORRECT** ANSWER IS: 1807

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152. QUESTION: Atomic number of potassium is \_\_\_\_\_.

- A. 17
- B. 18
- C. 19
- D. 20

**CORRECT** ANSWER IS: 19

153. QUESTION: Atomic symbol of Potassium is \_\_\_\_\_.

- A. P
- B. T
- C. K
- D. A

**CORRECT** ANSWER IS: K

154. QUESTION: Atomic weight of potassium is \_\_\_\_\_.

- A. 19.10
- B. 29.10
- C. 39.10
- D. 49.10

**CORRECT** ANSWER IS: 39.10

155. QUESTION: Electronic configuration of potassium is \_\_\_\_\_.

- A. 1s1
- B. 2s1
- C. 3s1
- D. 4s1

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**CORRECT** ANSWER IS: 4s1

156. QUESTION: Atomic radius of potassium is \_\_\_\_\_.

- A. 175 pm
- B. 275 pm
- C. 375 pm
- D. 475 pm

**CORRECT** ANSWER IS: 275 pm

157. QUESTION: Melting point of potassium is \_\_\_\_\_.

- A. 63.5 oC
- B. 64.5 oC
- C. 65.5 oC
- D. 66.5 oC

**CORRECT** ANSWER IS: 65.5 oC

158. QUESTION: Boiling Point of Potassium is \_\_\_\_\_.

- A. 559 oC
- B. 659 oC
- C. 759 oC
- D. 859 oC

**CORRECT** ANSWER IS: 759 oC

159. QUESTION: Oxidation states of potassium is \_\_\_\_\_.

- A. -1, +1 an amphoteric oxide
- B. +1 Strongly basic Oxide
- C. +2, +1 an amphoteric oxide
- D. +1, -1 a strongly basic oxide

**CORRECT** ANSWER IS: +1, -1 a strongly basic oxide

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160. QUESTION: Rubidium is discovered by \_\_\_\_\_

- A. Robert Bunsen
- B. Gustav Kirchhoff
- C. Both a & b
- D. None of the above

**CORRECT** ANSWER IS: Both a & b

161. QUESTION: Rubidium was discovered in \_\_\_\_\_.

- A. 1859
- B. 1860
- C. 1861
- D. 1862

**CORRECT** ANSWER IS: 181

162. QUESTION: Atomic Number of Rubidium is \_\_\_\_\_.

- A. 35
- B. 36
- C. 37
- D. 38

**CORRECT** ANSWER IS: 37

163. QUESTION: Atomic symbol of rubidium is \_\_\_\_\_.

- A. Rm
- B. Rb
- C. Ri
- D. Ru

**CORRECT** ANSWER IS: Rb

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164. QUESTION: Atomic weight of rubidium is \_\_\_\_\_.

- A. 82.47
- B. 83.47
- C. 84.47
- D. 85.47

**CORRECT** ANSWER IS: 85.47

165. QUESTION: Electronic configuration of rubidium is \_\_\_\_\_.

- A. 2s1
- B. 3s1
- C. 4s1
- D. 5s1

**CORRECT** ANSWER IS: 5s1

166. QUESTION: Atomic radius of rubidium is \_\_\_\_\_.

- A. 300 pm
- B. 301 pm
- C. 302 pm
- D. 303 pm

**CORRECT** ANSWER IS: 303 pm

167. QUESTION: Melting point of rubidium is \_\_\_\_\_.

- A. 35.3
- B. 36.3
- C. 37.3
- D. 39.3

**CORRECT** ANSWER IS: 39.3

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168. QUESTION: Boiling point of rubidium is \_\_\_\_\_.

- A. 685 oC
- B. 686 oC
- C. 687 oC
- D. 688 oC

**CORRECT** ANSWER IS: 688 oC

169. QUESTION: Oxidation states of rubidium is \_\_\_\_\_.

- A. -1, +1 an amphoteric oxide
- B. +1 Strongly basic Oxide
- C. +2, +1 an amphoteric oxide
- D. +1, -1 a strongly basic oxide

**CORRECT** ANSWER IS: +1, -1 a strongly basic oxide

170. QUESTION: Cesium was discovered by \_\_\_\_\_.

- A. Robert Bunsen
- B. Gustav Kirchhoff
- C. Both a & b
- D. None of the above

**CORRECT** ANSWER IS: Both a & b

171. QUESTION: Cesium was discovered in \_\_\_\_\_.

- A. 1660
- B. 1760
- C. 1860
- D. 1960

**CORRECT** ANSWER IS: 1860

172. QUESTION: Atomic number of cesium is \_\_\_\_\_.

- A. 22
- B. 33
- C. 44
- D. 55

**CORRECT** ANSWER IS: 55

173. QUESTION: Atomic symbol of cesium is \_\_\_\_\_.

- A. Cs
- B. Ci
- C. Cm
- D. All of the above

**CORRECT** ANSWER IS: Cs

174. QUESTION: Atomic weight of cesium is \_\_\_\_\_.

- A. 132.7
- B. 132.8
- C. 132.9
- D. 133.0

**CORRECT** ANSWER IS: 132.9

175. QUESTION: Electronic configuration of cesium is \_\_\_\_\_.

- A. 4s1
- B. 5s1
- C. 6s1

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D. 7s1

**CORRECT** ANSWER IS: 6s1

176. QUESTION: Atomic radius of cesium is \_\_\_\_\_,

A. 340 pm

B. 341 pm

C. 342 pm

D. 343 pm

**CORRECT** ANSWER IS: 343 pm

177. QUESTION: Melting point of cesium is \_\_\_\_\_ °C.

A. 28.2 °C

B. 28.3 °C

C. 28.4 °C

D. 28.5 °C

**CORRECT** ANSWER IS: 28.5 °C

178. QUESTION: Boiling point of cesium is \_\_\_\_\_ °C.

A. 670 °C

B. 671 °C

C. 672 °C

D. 673 °C

**CORRECT** ANSWER IS: 671 °C

179. QUESTION: Oxidation state of cesium is \_\_\_\_\_.

A. +1, -1

B. +2, +1

C. 1

D. -1

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**CORRECT** ANSWER IS: +1, -1

180. QUESTION: Francium was discovered by

\_\_\_\_\_.

- A. Marguerite Perey
- B. Carl Wilhelm Scheele
- C. Henry Cavendish
- D. All of the above

**CORRECT** ANSWER IS: Marguerite Perey

181. QUESTION: Atomic symbol of francium is \_\_\_\_\_.

- A. Fm
- B. Fr
- C. Fa
- D. Fc

**CORRECT** ANSWER IS: Fr

182. QUESTION: Atomic number of francium is

\_\_\_\_\_.

- A. 86
- B. 87
- C. 88
- D. 89

**CORRECT** ANSWER IS: 87

183. QUESTION: Atomic weight of francium is \_\_\_\_\_.

- A. 213

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B. 223

C. 233

D. 243

**CORRECT** ANSWER IS: 223

184. QUESTION: Electronic configuration of francium is

\_\_\_\_\_.

A. 4s1

B. 5s1

C. 6s1

D. 7s1

**CORRECT** ANSWER IS: 7s1

185. QUESTION: Atomic radius of francium is \_\_\_\_\_.

A. 248 pm

B. 348 pm

C. 448 pm

D. 558 pm

**CORRECT** ANSWER IS: 348 pm

186. QUESTION: Melting point of francium is \_\_\_\_\_.

A. 30 oC

B. 40 oC

C. 50 oC

D. 60 oC

**CORRECT** ANSWER IS: 30 oC

187. QUESTION: boiling point of francium is \_\_\_\_\_--.

A. 480 oC

B. 580 oC

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C. 680 oC

D. 780 oC

**CORRECT** ANSWER IS: 680 oC

188. QUESTION: oxidation state of francium is \_\_\_\_\_.

A. +1, -1

B. +2, +1

C. 1

D. +1

**CORRECT** ANSWER IS: 1

189. QUESTION: Francium was discovered in \_\_\_\_\_.

A. 1639

B. 1739

C. 1839

D. 1939

**CORRECT** ANSWER IS: 1939

190. QUESTION: Beryllium was discovered by \_\_\_\_\_.

A. Joseph Priestley

B. Carl Wilhelm Scheele

C. Louis Nicolas Vauquelin

D. None of the above

**CORRECT** ANSWER IS: Louis Nicolas Vauquelin

191. QUESTION: Beryllium was discovered in \_\_\_\_\_.

A. 1798

B. 1898

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C. 1998

D. 2008

**CORRECT** ANSWER IS: 1798

192. QUESTION: Atomic number of beryllium is \_\_\_\_\_.

A. 1

B. 2

C. 3

D. 4

**CORRECT** ANSWER IS: 4

193. QUESTION: Atomic symbol of beryllium is \_\_\_\_\_.

A. Br

B. Be

C. Bm

D. BI

**CORRECT** ANSWER IS: Be

194. QUESTION: Atomic weight of beryllium is \_\_\_\_\_.

A. 9.012

B. 9.12

C. 9.021

D. 9.21

**CORRECT** ANSWER IS: 9.012

195. QUESTION: Electronic configuration of beryllium is

\_\_\_\_\_.

A. 1s<sup>2</sup>

B. 2s<sup>2</sup>

C. 3s<sup>2</sup>

D. 4s<sup>2</sup>

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**CORRECT** ANSWER IS: 2s2

196. QUESTION: Atomic radius of beryllium is \_\_\_\_\_.

- A. 150 pm
- B. 151 pm
- C. 152 pm
- D. 153 pm

**CORRECT** ANSWER IS: 153 pm

197. QUESTION: Melting point of beryllium is \_\_\_\_\_.

- A. 1087 oC
- B. 1187 oC
- C. 1287 oC
- D. 1387 oC

**CORRECT** ANSWER IS: 1287 oC

198. QUESTION: Boiling point of beryllium is \_\_\_\_\_.

- A. 2369 oC
- B. 2469 oC
- C. 2569 oC
- D. 2669 oC

**CORRECT** ANSWER IS: 2469 oC

199. QUESTION: Oxidation state of beryllium is \_\_\_\_\_.

- A. +1, -1
- B. +2, +1
- C. +1
- D. 1

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**CORRECT** ANSWER IS: +2, +1

200. QUESTION: Magnesium was discovered by

\_\_\_\_\_.

- A. Robert Bunsen
- B. Carl Wilhelm Scheele
- C. Joseph Black
- D. All of the above

**CORRECT** ANSWER IS: Joseph Black

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